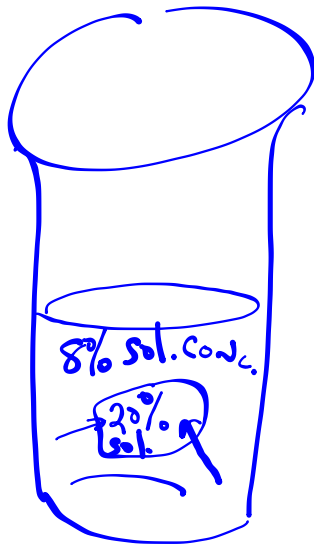


If conc. of solute in ENV.  
is  $>$  inside cell

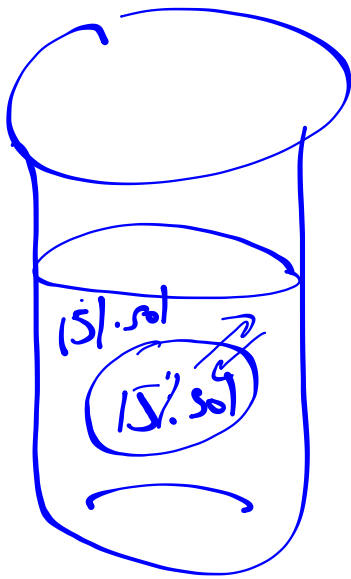
= HyperTonic  
solution.

$H_2O$  conc. is lower outside,  
So MOVES out



If conc. of solute in  
Environment is  $<$   
conc. inside cell  
= Hypotonic  
Solution

H<sub>2</sub>O conc. is higher  
outside, so moves in



If concentration of  
Solute in environment.  
is  $\ominus$  to conc. inside  
= Isotonic.

$H_2O$  moves in/out  
Equally.