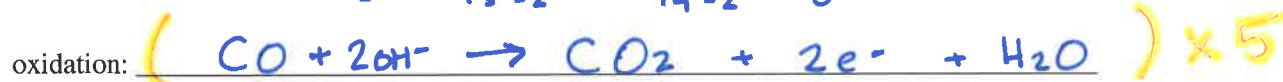
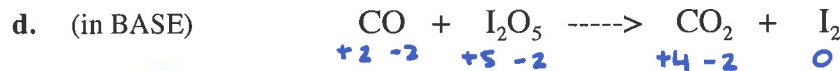
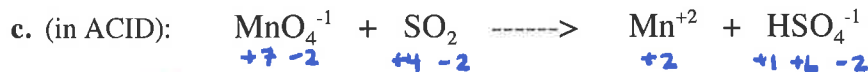
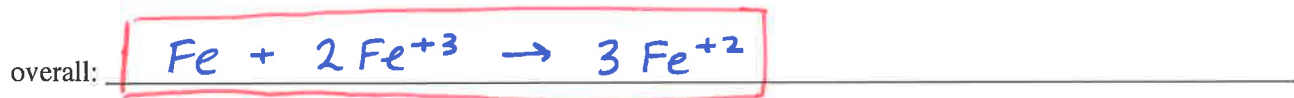
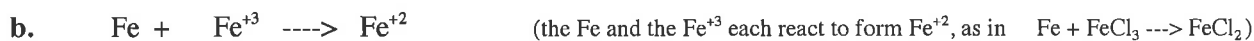
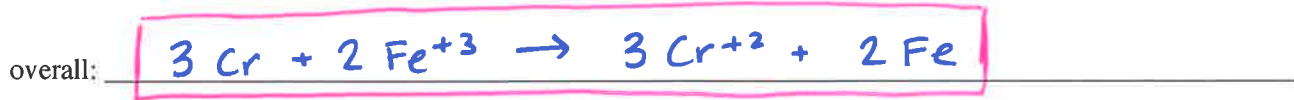
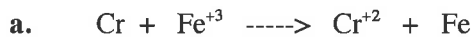
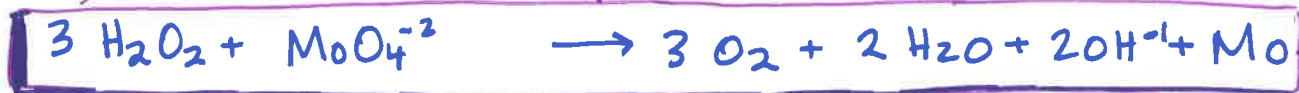
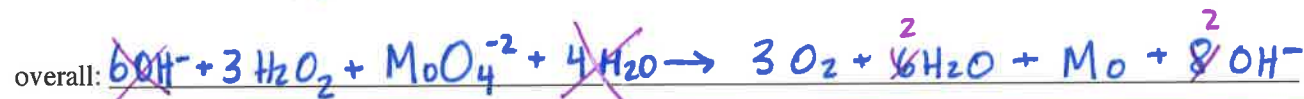
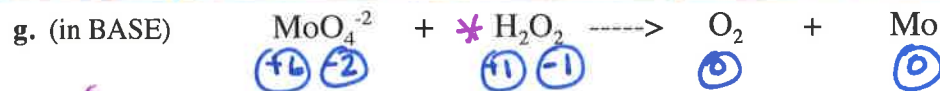
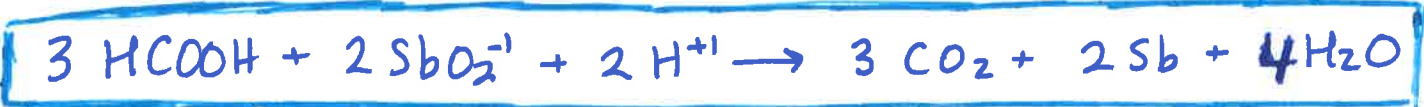
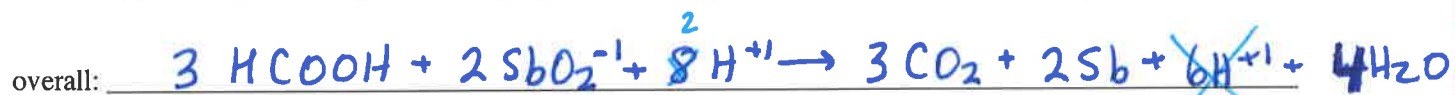
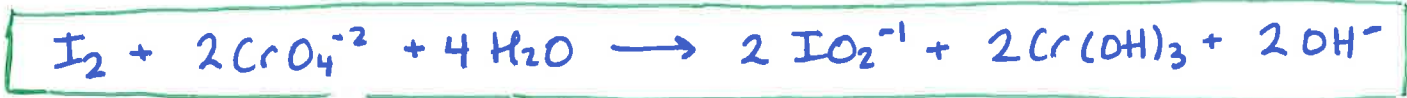
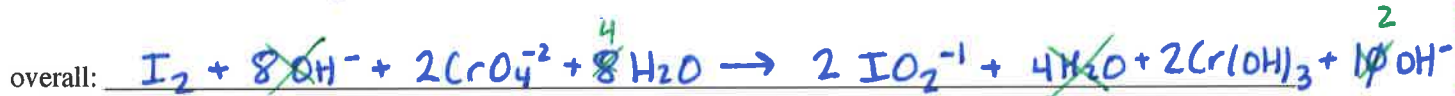
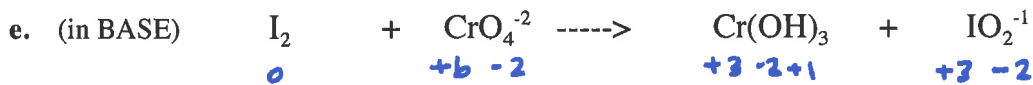


Problem X. (Optional review for the redox balancing quiz!)

For each reaction, Write the balanced oxidation half reaction.
Write the balanced reduction half reaction.
Write the overall balanced redox reaction.





* this is a peroxide so oxygen is (-1)

