

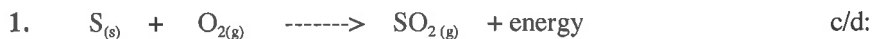
Each chemical equation below shows two elements reacting to form a compound. For each reaction:

a) classify the compound that forms as ionic or covalent.

b) name the compound that forms.

c) state whether the each reacting element will need to gain, lose, or share electrons to bond in the compound.

d) If electrons are lost or gained, indicate which element will gain electrons and which will lose electrons.



a. _____

b. _____



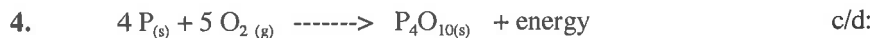
a. _____

b. _____



a. _____

b. _____



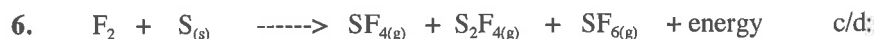
a. _____

b. _____



a. _____

b. _____



a. _____

b. _____



a. _____

b. _____



a. _____

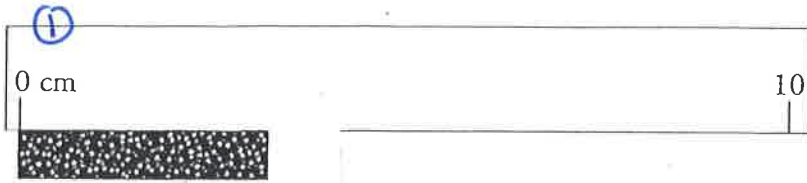
b. _____



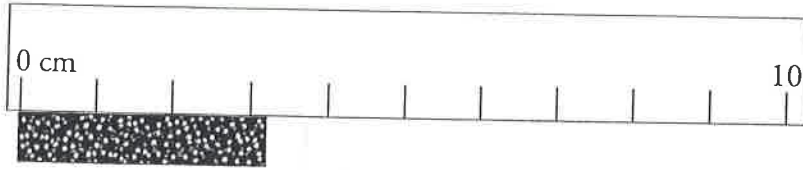
a. _____

b. _____

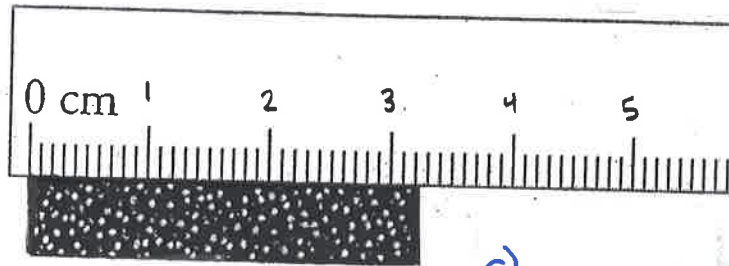
WS 2.0 Part I: Reporting data to the correct number of significant figures!



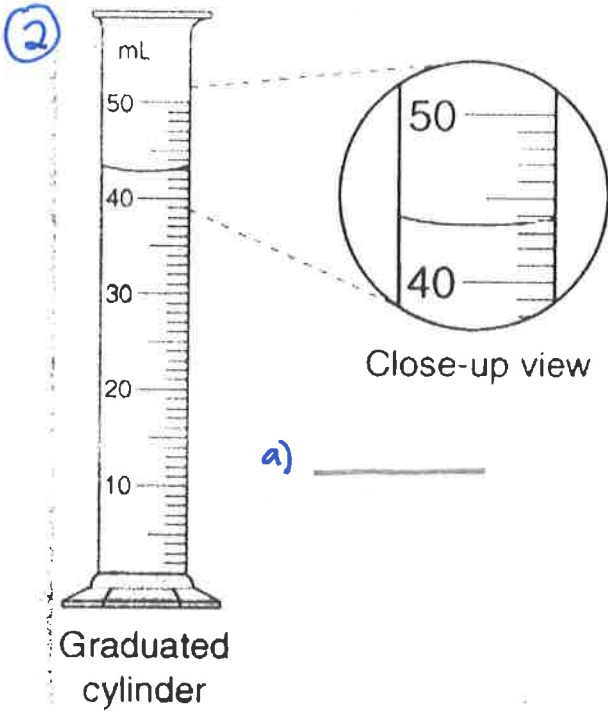
a) _____



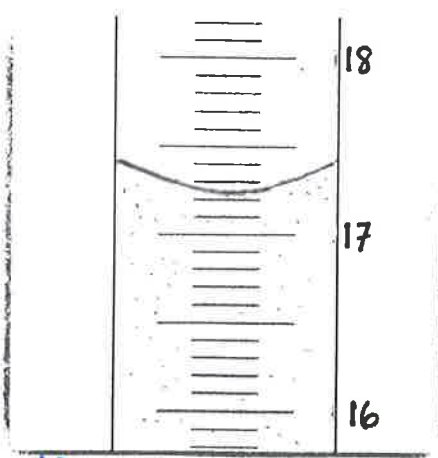
b) _____



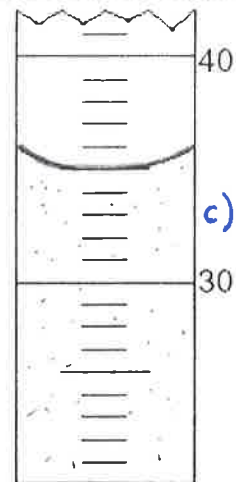
c) _____



a) _____



b) _____



c) _____

